ENANTIOSELECTIVE SYNTHESIS OF (-)-METHYL 5-LACTYLSHIKIMATE LACTONE

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<u>Abstract</u> -- A short synthesis of enantiomerically pure (-)-methyl 5-lactylshikimate lactone 1 from (-)-shikimic acid is described, which, along with NOE measurements, establishes its absolute stereostructure.

In the course of identifying and screening new fungal metabolites for biological activity, Isogai <u>et al</u>. recently reported a new shikimate-derived metabolite 1 isolated from *Penicillium* sp. K-114.¹ The structure of lactone 1 is noteworthy for several reasons. It represents the first lactone (and first O-lactyl derivative) isolated in the shikimate/chorismate pathway.^{2,3} Its biosynthesis is also intriguing, since 1 might originate along the main stem of the pathway from 5-enolpyruvylshikimate-3-phosphate 2 or from the dead-end metabolite 5enolpyruvylshikimic acid 3 (Compound Z-1).⁴ In either case, reduction of the enolpyruvyl double bond is required as in the formation of UDP-N-acetylmuramic acid, an intermediate in bacterial cell wall peptidoglycan synthesis.⁵ Here we disclose a short, enantioselective synthesis of 1 confirming its absolute stereostructure as shown. The lactyl ether appendage is attached by a diazo coupling reaction whose mild conditions are compatible with both acid-sensitive and electrophilic groups.



Treating the known acetonide 4 of methyl shikimate⁶ with ethyl diazopropionate (5 equiv, catalytic HBF₄, CH₂Cl₂, 0°C) gave diester 5 in 25% yield as a mixture of C-8 diastereomers.⁷ Since the direct conversion of 5 to 1 failed using various solvent/acid combinations, the acetonide protecting group in 5 was first hydrolyzed (4:1 HOAc:H₂O, 75°C, 6 h) to afford 6 (83%). When both diastereomers of 6 were suspended in dry benzene and boiled for 10 h with a catalytic amount of pyridinium *p*-toluenesulfonate,⁸ lactonization and C-8 equilibration led to methyl 5-lactylshikimate lactone 1 which was obtained as a single isomer after flash chromatography: (68%, mp 128-129°C, lit¹ mp 133-135°C). Spectral, analytical and chiroptical data for synthetic 1 matched values reported for the natural product. Moreover a nuclear Overhauser effect (5%) between H5 and H8 confirmed the equatorial C8-methyl in 1 and established the overall 3R,4R,5R,8S-configuration.⁹



REFERENCES AND NOTES

- 1. Isogai, A.; Washizu, M.; Murakoshi, S.; Suzuki, A. Agric. Biol. Chem. 1985, 49, 167.
- 2. (a) Ganem, B. Tetrahedron 1978, <u>34</u>, 3353.
- Although no naturally-occurring lactones have heretofore been reported, lactone formation has been noted on numerous occasions in the chemistry of shikimate pathway derivatives: see (a) Ganem, B.; Ikota, N. J. Chem. Soc. Chem. Commun. 1978, 869; (b) Teng, C.-Y. P.; Ganem, B. Tetrahedron Lett. 1985, 26, 21. (c) Refs. 4a,b.
- (a) Davis, B.D.; Mingioli, E.S. J. Bacteriol. 1953, <u>66</u>, 129. (b) McGowan, D.A.; Berchtold, G.A. J. Am. Chem. Soc. 1982, <u>104</u>, 7036.
- 5. Zemell, R.I.; Anwar, R.A. J. Biol. Chem. 1975, 250, 3185.
- 6. Diels, O.; Fritsche, P. <u>Chem. Ber</u>. 1911, <u>44</u>, 3018.
- 7. Satisfactory ¹H-NMR, IR and HRMS data were obtained for this and all other new compounds.
- 8. Miyashita, M.; Yoshikoshi, A.; Grieco, P.A. J. Org. Chem. 1977, 42, 3772.
- 9. Generous financial support from the National Institutes of Health (GM 24054) as well as support of the Cornell Nuclear Magnetic Resonance Facility by the NSF (CHE 7904825; PGM 8018643) and NIH (RR02002) is gratefully acknowledged. We thank Dr. Aidan Harrison of that facility for help with NOE measurements.

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